

Chemistry Review: Formula Writing & Nomenclature Name _____

Write the formulas or names for the following:

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|--|-------|------------------------------------|-------|
| 1. lithium sulfide | _____ | 26. carbon dioxide | _____ |
| 2. beryllium fluoride | _____ | 27. oxygen difluoride | _____ |
| 3. magnesium oxide | _____ | 28. carbon tetrachloride | _____ |
| 4. cesium phosphide | _____ | 29. SO ₃ | _____ |
| 5. barium bromide | _____ | 30. N ₂ O ₅ | _____ |
| 6. BaI ₂ | _____ | 31. C ₁₀ N ₆ | _____ |
| 7. K ₂ O | _____ | 32. heptaoxygen tetrabromide | _____ |
| 8. Li ₃ P | _____ | 33. triphosphorus octasulfide | _____ |
| 9. NaBrO ₃ | _____ | 34. nitric acid | _____ |
| 10. CaCr ₂ O ₇ | _____ | 35. dinitrogen monofluoride | _____ |
| 11. magnesium chlorite | _____ | 36. HCl | _____ |
| 12. aluminum sulfide | _____ | 37. H ₂ SO ₄ | _____ |
| 13. barium hypochlorite | _____ | 38. ammonium nitrate | _____ |
| 14. sodium sulfate | _____ | 39. KOH | _____ |
| 15. ammonium carbonate | _____ | 40. H ₃ PO ₄ | _____ |
| 16. (NH ₄) ₂ CrO ₄ | _____ | 41. acetic acid | _____ |
| 17. Al ₂ (SO ₄) ₃ | _____ | 42. carbonic acid | _____ |
| 18. Be(NO ₃) ₂ | _____ | 43. sulfuric acid | _____ |
| 19. K ₂ CO ₃ | _____ | 44. copper (II) oxide | _____ |
| 20. Mg(NO ₂) ₂ | _____ | 45. iron (III) nitrite | _____ |
| 21. iron (II) iodide | _____ | 46. tin IV acetate | _____ |
| 22. lead (II) sulfite | _____ | 47. lead II carbonate | _____ |
| 23. copper (I) sulfide | _____ | 48. tin (II) fluoride | _____ |
| 24. Ni ₃ (PO ₄) ₂ | _____ | 49. sodium bicarbonate | _____ |
| 25. Cr(OH) ₃ | _____ | 50. hydrogen sulfide | _____ |