1. What is the molarity of carbonic acid if 25.0 mL of the solution is neutralized by 48.3 mL of 0.20 M NaOH?
Write the balanced equation:

2. What is the molarity of sodium hydroxide if 20.0 mL of the solution is neutralized by 17.4 mL of 1.0 M H₃PO₄?
Write the balanced equation:

3. Calculate the molarity of an HCl solution if 25.0 mL of the solution is neutralized by 15.5 mL of 0.800M NaOH.
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