

Chemistry
 Covalent Bonding II – Lewis, VSEPR, Polarity, Intermolecular Attractions

Name _____
 Date _____

<u>Name</u>	<u>Formula</u>	<u>Lewis Dot Structure</u>	<u>Molecular Shape</u> <i>VSEPR Shape</i>	<u>Bond Polarity</u> <i>Use EN differences to calculate</i>	<u>Molecule Type</u> <i>polar or nonpolar based on molecule symmetry</i>	<u>Intermolecular Attractions</u> <i>London dispersion, dipole, hydrogen bonding?</i>
Nitrogen trichloride						
	CO					
Ammonium						
	NO ₃ ⁻					
Fluorine gas						
N/A	COBr ₂					
Water						
	SF ₆					