

Name: _____

Date: _____

Nomenclature - 180 Names

1. Na_2S
2. Li_2O
3. MgBr_2
4. Cl_2O
5. NO
6. H_2S
7. HCl
8. AlN
9. CaF_2
10. KI
11. $\text{Mg}(\text{OH})_2$
12. NaOH
13. CaO
14. Na_2SO_4
15. NaNO_3
16. $\text{Mg}(\text{NO}_3)_2$
17. KClO_4
18. NH_4ClO_3
19. BaCO_3
20. KNO_2
21. MgSO_4
22. $\text{Ca}(\text{ClO})_2$
23. $\text{Ba}(\text{NO}_2)_2$
24. $\text{Ba}(\text{ClO}_3)_2$
25. CaSO_3
26. KBrO_3
27. $\text{Cd}(\text{IO}_3)_2$
28. NaHSO_4
29. KNaCO_3
30. $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$
31. Lithium fluoride
32. Lithium chloride
33. Lithium bromide
34. Lithium oxide
35. Lithium sulfide
36. Sodium fluoride
37. Sodium chloride
38. Sodium bromide
39. Sodium iodide
40. Sodium oxide
41. Sodium sulfide
42. Potassium fluoride
43. Potassium chloride
44. Potassium bromide
45. Potassium iodide
46. Potassium oxide
47. Ammonium fluoride
48. Ammonium chloride
49. Ammonium bromide
50. Ammonium iodide
51. Ammonium sulfide
52. Magnesium chloride
53. Magnesium bromide
54. Magnesium hydroxide
55. Magnesium sulfide
56. Calcium chloride
57. Calcium bromide
58. Calcium hydroxide
59. Calcium oxide
60. Calcium cyanide

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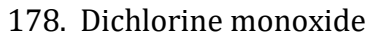
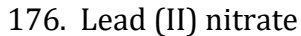
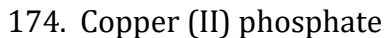
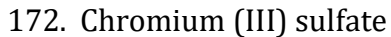
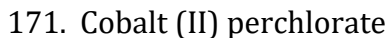
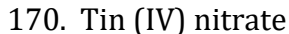
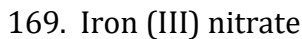
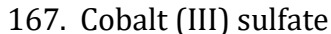
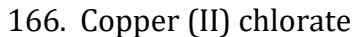
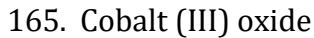
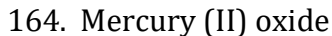
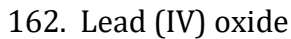
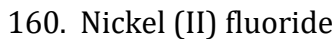
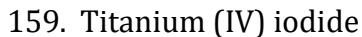
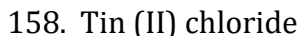
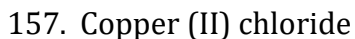
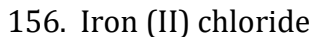
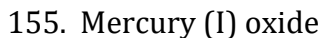
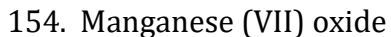
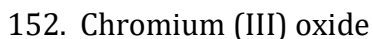
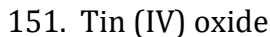
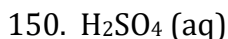
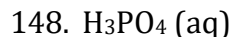
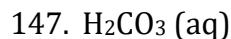
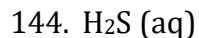
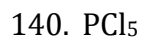
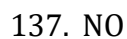
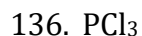
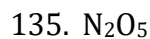
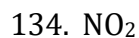
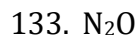
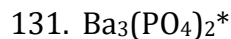
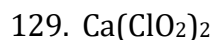
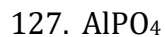
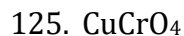
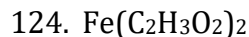
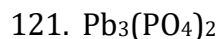
Nomenclature - 180 Names

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|------------------------------|--|
| 61. Calcium sulfide | 91. TiCl_2 |
| 62. Sodium hydroxide | 92. TiBr_4 |
| 63. Sodium cyanide | 93. CuCl |
| 64. Ammonium cyanide | 94. PbI_2 |
| 65. Barium chloride | 95. SnCl_4 |
| 66. Barium hydroxide | 96. Cu_2O |
| 67. Barium cyanide | 97. CrO_3 |
| 68. Barium oxide | 98. Mn_2O_3 |
| 69. Barium sulfide | 99. TiO_2 |
| 70. Calcium sulfate | 100. PbO |
| 71. Sodium nitrate | 101. TiCl_3 |
| 72. Potassium perchlorate | 102. NiBr_2 |
| 73. Aluminum sulfate | 103. CuBr_2 |
| 74. Potassium chlorate | 104. PbCl_2 |
| 75. Sodium hypochlorite | 105. CrF_3 |
| 76. Barium phosphate | 106. CuO |
| 77. Ammonium carbonate | 107. SnO |
| 78. Magnesium sulfite | 108. TiO |
| 79. Lithium nitrite | 109. Fe_2O_3 |
| 80. Sodium chlorite | 110. Mn_2O_3 |
| 81. Ammonium dichromate | 111. $\text{Fe}_2(\text{SO}_4)_3$ |
| 82. Sodium nitrite | 112. $\text{Cr}(\text{OH})_2$ |
| 83. Sodium bromate | 113. HgClO_3 |
| 84. Potassium phosphate | 114. $\text{Fe}(\text{ClO}_4)_2$ |
| 85. Silver carbonate | 115. MnSO_4 |
| 86. Manganese (III) chloride | 116. $\text{Cr}_2(\text{SO}_3)_3$ |
| 87. Iron (III) bromide | 117. $\text{Hg}(\text{IO}_3)_2$ |
| 88. Chromium (III) bromide | 118. $\text{Pb}(\text{ClO}_2)_2$ |
| 89. Tin (IV) chloride | 119. $\text{Cu}(\text{C}_2\text{H}_3\text{O}_2)_2$ |
| 90. Manganese (II) bromide | 120. Cu_2SO_4 |

Name: _____

Date: _____

Nomenclature - 180 Names



*The original 180 Names document, created on a typewriter, had two #131s. This updated version keeps the tradition.