

KEY

Chapter 3 Review Questions: Dimensional Analysis

HONORS ONLY

- 1. Racing cars at the Indianapolis Motor Speedway routinely travel around the track at an average speed of 0.0963 km/sec. What is this speed in miles/hour? (1 km = 0.62 mi)

$$\frac{0.0963 \text{ km}}{\text{sec}} \times \frac{0.62 \text{ mi}}{1 \text{ km}} \times \frac{3600 \text{ sec}}{1 \text{ hr}} = 215 \text{ mi/hr}$$

- 2. The length of a marathon race is approximately 26.2 miles. What is the distance in kilometers? (1 km = 0.62 miles)

$$\frac{26.2 \text{ mi}}{1} \times \frac{1 \text{ km}}{0.62 \text{ mi}} = 42.26 \text{ km}$$

- 3. A tank contains 5,000 L of chlorine gas. Express this volume in cubic centimeters.

$$\frac{5,000 \text{ L}}{1} \times \frac{1000 \text{ mL}}{1 \text{ L}} \times \frac{1 \text{ cm}^3}{1 \text{ mL}} = 5 \times 10^6 \text{ cm}^3$$

- 4. Light travels at a speed of 186,000 miles/second. How many kilometers will it travel in 2 hours? (1 km = 0.6 miles)

$$\frac{186,000 \text{ mi}}{\text{sec}} \times \frac{1 \text{ km}}{0.62 \text{ mi}} \times \frac{3600 \text{ sec}}{1 \text{ hr}} \times \frac{2 \text{ hr}}{1} = 2.16 \times 10^9 \text{ km}$$

- 5. Stephen weighs 59,000 grams and is 5'11." Will he be a likely recruit for the offensive line on the Penn State football team? (1 kg = 2.2 lbs)

$$\frac{59,000 \text{ g}}{1} \times \frac{1 \text{ kg}}{1000 \text{ g}} \times \frac{2.2 \text{ lbs}}{1 \text{ kg}} = 129.8 \text{ lbs} \quad \text{NO}$$

- 6. What is the volume in Liters of a swimming pool that has the dimensions of 10 cm x 25 m x 12.9 mm?

$$\frac{25 \text{ m}}{1} \times \frac{100 \text{ cm}}{1 \text{ m}} = 2,500 \text{ cm} \quad \frac{12.9 \text{ mm}}{1} \times \frac{1 \text{ cm}}{10 \text{ mm}} = 1.29 \text{ cm} \quad V = 10 \text{ cm} \times 2500 \text{ cm} \times 1.29 \text{ cm} = 32,250 \text{ cm}^3$$

$$32,250 \text{ cm}^3 \times \frac{1 \text{ mL}}{1 \text{ cm}^3} \times \frac{1 \text{ L}}{1000 \text{ mL}} = 32.25 \text{ L}$$

- 7. A student wants to identify the main component in a commercial liquid cleaner. He finds that 35.8 mL of the cleaner weighs 28.1 g. Of the following possibilities, which is the main component of the cleaner?

Substance	Density, g/cm ³
Chloroform	1.483
Diethyl ether	0.714
<u>Isopropyl alcohol</u>	0.785
Toluene	0.867

$$D = \frac{28.1 \text{ g}}{35.8 \text{ mL}} = 0.785 \text{ g/mL}$$

- 8. For a material to float on the surface of water, the material must have a density of less than that of water (1.0 g/mL) and must not react with the water or dissolve in it. A spherical ball has a radius of 0.5 cm and weighs 2.0 g. Will this ball float or sink when placed in water? (Volume of a sphere = 4/3[πr³])

$$V = \frac{4}{3} \pi (0.5 \text{ cm})^3 = 0.524 \text{ cm}^3$$

$$D = \frac{2.0 \text{ g}}{0.524 \text{ cm}^3} = 3.82 \text{ g/cm}^3 \quad \text{SINK}$$

- 9. Perform the following conversions:

a. 8.43 cm to millimeters

$$\frac{8.43 \text{ cm}}{1} \times \frac{10 \text{ mm}}{1 \text{ cm}} = 84.3 \text{ mm}$$

b. 2.41 x 10² cm to meters

$$\frac{2.41 \times 10^2 \text{ cm}}{1} \times \frac{1 \text{ m}}{100 \text{ cm}} = 2.41 \text{ m}$$

c. 1.445 x 10⁵ mm to kilometers

$$\frac{1.445 \times 10^5 \text{ mm}}{1} \times \frac{1 \text{ km}}{10^6 \text{ mm}} = 0.1445 \text{ km}$$

d. 550 cm³ of water to Liters

$$\frac{550 \text{ cm}^3}{1} \times \frac{1 \text{ mL}}{1 \text{ cm}^3} \times \frac{1 \text{ L}}{1000 \text{ mL}} = 0.55 \text{ L}$$

- 10. If a bee flies at an average speed of 3.4 m/s, what is its average speed in miles/hour? (1 km = 0.62 miles)

$$\frac{3.4 \text{ m}}{\text{sec}} \times \frac{1 \text{ km}}{1000 \text{ m}} \times \frac{0.62 \text{ mi}}{1 \text{ km}} \times \frac{3600 \text{ sec}}{1 \text{ hr}} = 7.59 \text{ mi/hr}$$

- 11. Mercury is traded by the "flask," a unit that has a mass of 34.5 kg. What is the volume of a "flask" of mercury if the density of mercury is 13.6 g/mL?

$$D = \frac{M}{V} \quad \frac{34.5 \text{ kg}}{1} \times \frac{1000 \text{ g}}{1 \text{ kg}} = 34,500 \text{ g}$$

$$V = \frac{M}{D} \quad V = \frac{34,500 \text{ g}}{13.6 \text{ g/mL}} = 2536.8 \text{ mL}$$